

## Solutions to assignment 7

1. (a) The current density is  $j = I/A = I/\pi(d/2)^2$  and has the value

$$j = \frac{4(10^{-3} \text{ A})}{\pi(1.63 \times 10^{-3} \text{ m})^2} = 479 \text{ A m}^{-2}.$$

- (b) The drift velocity is  $v_d = I/An_e = j/ne$ . Using the  $j$  value from part (a), we find

$$\begin{aligned} v_d &= \frac{479 \text{ A m}^{-2}}{(8.47 \times 10^{28} \text{ m}^{-3})(1.602 \times 10^{-10} \text{ C})} \\ &= 3.53 \times 10^{-8} \text{ m/s} \\ &= 35.3 \text{ nm/s}. \end{aligned}$$

2. The Fermi speed is found by inverting  $E_F = \frac{1}{2}mu_F^2$ . That is,

$$u_F = \left(\frac{2E_F}{m_e}\right)^{1/2} = c\left(\frac{2E_F}{m_e c^2}\right)^{1/2},$$

where  $c$  is the speed of light.

- (a) For Na:

$$\begin{aligned} u_F &= (3.00 \times 10^8 \text{ m/s})\sqrt{\frac{2(3.26 \text{ eV})}{5.11 \times 10^5 \text{ eV}}} \\ &= 1.07 \times 10^6 \text{ m/s}. \end{aligned}$$

- (b) For Au:

$$\begin{aligned} u_F &= (3.00 \times 10^8 \text{ m/s})\sqrt{\frac{2(5.55 \text{ eV})}{5.11 \times 10^5 \text{ eV}}} \\ &= 1.40 \times 10^6 \text{ m/s}. \end{aligned}$$

- (c) For Sn:

$$\begin{aligned} u_F &= (3.00 \times 10^8 \text{ m/s})\sqrt{\frac{2(10.3 \text{ eV})}{5.11 \times 10^5 \text{ eV}}} \\ &= 1.90 \times 10^6 \text{ m/s}. \end{aligned}$$

3. The resistivity is given by  $\rho = m_e u_F / ne^2 \lambda$ , where  $n$  is the electronic density and  $\lambda$  is the mean free path. The electronic densities of Na, Au, and Sb are listed in Table 10-3 on page 458 of the text. The corresponding Fermi velocities are known from question 2. The corresponding resistivities are  $4.2 \mu\Omega \text{ cm}$ ,  $2.04 \mu\Omega \text{ cm}$ , and  $10.6 \mu\Omega \text{ cm}$ . Hence,

$$\begin{aligned} \lambda_{\text{Na}} &= \frac{m_e u_F}{ne^2 \rho} \\ &= \frac{(9.109 \times 10^{-31} \text{ kg})(1.07 \times 10^6 \text{ m/s})}{(2.65 \times 10^{28} \text{ m}^{-3})(1.602 \times 10^{-19} \text{ C})^2(4.2 \times 10^{-8} \Omega \text{ m})} \\ &= 3.4 \times 10^{-8} \text{ m} \\ &= 34 \text{ nm} \end{aligned}$$

$$\begin{aligned} \lambda_{\text{Au}} &= \frac{(9.109 \times 10^{-31} \text{ kg})(1.40 \times 10^6 \text{ m/s})}{(5.9 \times 10^{28} \text{ m}^{-3})(1.602 \times 10^{-19} \text{ C})^2(2.04 \times 10^{-8} \Omega \text{ m})} \\ &= 4.1 \times 10^{-8} \text{ m} \\ &= 41 \text{ nm} \end{aligned}$$

$$\begin{aligned} \lambda_{\text{Sn}} &= \frac{(9.109 \times 10^{-31} \text{ kg})(1.90 \times 10^6 \text{ m/s})}{(14.8 \times 10^{28} \text{ m}^{-3})(1.602 \times 10^{-19} \text{ C})^2(10.6 \times 10^{-8} \Omega \text{ m})} \\ &= 4.3 \times 10^{-9} \text{ m} \\ &= 4.3 \text{ nm} \end{aligned}$$

4. (a) The photon energy is related to its wavelength by  $E = hc/\lambda$ . Hence,

$$E = \frac{1240 \text{ eV nm}}{3.35 \times 10^3 \text{ nm}} = 0.37 \text{ eV}.$$

- (b) The corresponding temperature scale is

$$T = \frac{E}{k_B} = \frac{0.37 \text{ eV}}{8.617 \times 10^{-5} \text{ eV/K}} = 4300 \text{ K}.$$

5. (a) The energy as a function of the radial quantum number  $n$  can be organized as follows:

$$E_n = \underbrace{-\frac{m_e}{2} \left(\frac{e^2}{4\pi\epsilon_0 \hbar}\right)^2}_{-13.6 \text{ eV}} \frac{m^*}{m_e} \frac{1}{\kappa^2} \frac{1}{n^2}.$$

Therefore, the ionization energy is

$$E_\infty - E_1 = \frac{0.2(13.6 \text{ eV})}{(11.8)^2} = 0.0195 \text{ eV}.$$

- (b) The average orbital size is

$$\langle r_n \rangle = a_0 n^2 \frac{m_e}{m^*} \kappa,$$

where  $a_0 = 0.0529 \text{ nm}$  is the Bohr radius. Hence,

$$\langle r_1 \rangle = (0.0529 \text{ nm})(1)^2(1/0.2)(11.8) = 3.12 \text{ nm}.$$

- (c) For silicon,  $E_g = 1.11 \text{ eV}$  at room temperature. Thus,  $E_1/E_g = 0.0195/1.11 = 0.0176$  or about 2%.